NODERN DEVELOPMENT OF MAGNETIC RESONANCE

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Diluted Iron Oxide in K₂O-Al₂O₃-B₂O₃ Studied Method EPR

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Great interest in borate glasses and crystals is due to attractive physical properties. In particular, the borate compounds, un-doped and doped with rare-earth and transition elements, are very promising materials for nonlinear optics, quantum electronics and laser technology [1, 2], scintillators and thermosluminescent dosimeters [3], detectors and transformers of the ionizing radiation [4]. We investigated the $K_2O-Al_2O - B_2O_3$ (KAB) glasses. The first sample was doped



Fig. 1. The ESR spectra in KAB glasses with Fe₂O₃: a 2% and b 4%, before and after annealing.



Fig. 2. Mössbauer spectra in KAB: a before and b after annealing.

by 2% and second sample has 4% Fe_2O_3 iron oxide, the third sample KAB + 4% Fe_2O_3 was annealed 2 hours at 525 °C.

The continuous wave (CW) EPR spectra of iron oxides 2% and 4% before and after annealing were recorded on a Bruker EMX+ spectrometer at the frequency of 9.4 GHz. Three group of signals were observed in magnetic resonance spectra (see Fig. 1). The first line has $g \sim 4.2$. Detailed description of the features of the spin Hamiltonian for the observation of this line are given in the paper [5]. The magnetic resonance lines of complex shapes near $g_{\parallel} \sim 6$ and $g_{\perp} \sim 2$ and with weak intensity were observed in spectra samples before annealing. The line with $g_{\parallel} \sim 6$ was described by transition between level of doublet $|\pm 1/2\rangle$ for Fe³⁺ in octahedron position where term of ground state is ${}^{6}S_{5/2}$. The intensity of magnetic resonance signals with $g_{\perp} \sim 2$ and $g_{\parallel} \sim 8.2$ were increased in samples after annealing process. These lines can be connected with ferromagnetic clusters α -Fe₂O₃ of large size with strong magnetic anisotropy.

A Mössbauer spectra obtained with the Mössbauer spectrometer MS-1104EM at 300 K (see Fig. 2). The Mössbauer spectrum the sample KAB + 4%Fe₂O₃ before annealing at room temperature, is a set of 3 main doublets.

Probably the doublet the No. 1 relates to the atoms of Fe^{3+} in tetrahedral positions. The doublet No. 2 can be attributed to atoms Fe^{3+} in octahedron positions. Probably Fe^{2+} ions were occupied tetrahedral positions for a doublet number 3. After annealing, the Mössbauer spectrum has changed dramatically. The sextets were formed in third sample after annealing. We believe that ferromagnetic clusters (α -Fe₂O₃) with different size were formed in third sample.

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