

# ZEOLITES AS CARRIERS OF ANTITUMOR RIBONUCLEASE BINASE

Vera Khodzhaeva<sup>1</sup>, Oleg Lopatin<sup>2, 1</sup>, Pavel Zelenikhin<sup>1\*</sup>, Olga N. Ilinskaya<sup>1</sup>

<sup>1</sup>Department of Microbiology, Kazan Federal University, Russia, <sup>2</sup> Department of Mineralogy & Lithology, Kazan Federal University, Russia

Submitted to Journal: Frontiers in Pharmacology

Specialty Section: Translational Pharmacology

Article type: Original Research Article

Manuscript ID: 445483

Received on: 27 Dec 2018

Revised on: 02 Apr 2019

Frontiers website link: www.frontiersin.org



### Conflict of interest statement

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest

### Author contribution statement

VK, OL and OI planned experiments. VK and PZ performed experiments. OL and OI analyzed data. VK and OI wrote the paper.

### Keywords

Zeolites, Chabazite, Clinoptiolite, natrolite, Cytotoxicity, Cytotoxic RNAse, binase, Immobilization

### Abstract

### Word count: 176

Natural and synthetic zeolites have many applications in biomedicine and nutrition. Due to its properties, zeolites can absorb therapeutically active proteins and release them under physiological conditions. In this study we tested the clinoptilolite, chabazite and natrolite ability to be loaded by antitumor ribonuclease binase and the cytotoxicity of the complexes obtained. We found the optimal conditions for binase loading into zeolites and established the dynamic of its release. Cytotoxic effects of zeolite-binase complexes towards colorectal cancer CaCo2 cells were characterized after 24h and 48h of incubation with cells using MTT-test. Zeolites were toxic itself and reduced cells viability by 30% (clinoptilolite), 40% (chabazite) and 70% (natrolite) after 48 h of incubation but complexes of clinoptilolite as well as chabazite with binase demonstrated always enhanced toxicity (up to 57% and 60% for clinoptilolite and chabazite, respectively) in comparison to binase and zeolites separately. Our results contribute to the perspective development of binase-based complexes for therapy of colorectal cancer for or the treatment of malignant skin neoplasms when a complexes can be used in pasty form.

### Funding statement

The study was performed within the Russian Government Program of Competitive Growth of Kazan Federal University and was supported by the Russian Foundation for Basic Research (project no. 17-00-00060).

### Ethics statements

(Authors are required to state the ethical considerations of their study in the manuscript, including for cases where the study was exempt from ethical approval procedures)

Does the study presented in the manuscript involve human or animal subjects: No

### Data availability statement

Generated Statement: All datasets generated for this study are included in the manuscript and the supplementary files.

### ZEOLITES AS CARRIERS OF ANTITUMOR RIBONUCLEASE 1 **BINASE** 2

# Vera Khojaewa<sup>1</sup>, Oleg Lopatin<sup>2</sup>, Pavel Zelenikhin<sup>1\*</sup>, Olga Ilinskaya<sup>1</sup>

- <sup>1</sup>Institute of Fundamental Medicine and Biology, Department of Microbiology, Kazan Federal 5
- University, Kazan, Russia 6
- 7 <sup>2</sup>Institute of Geology and Oil &Gas Technologies, Department of Mineralogy & Lithology
- Kazan Federal University, Kazan, Russia 8

### 9 \* Correspondence:

- Dr. Pavel Zelenikhin 10
- pasha mic@mail.ru 11

### Keywords: zeolites, chabazite, clinoptilolite, natrolite, cytotoxic RNAse, binase, 12

13 immobilization

### Abstract 14

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Natural and synthetic zeolites have many applications in biomedicine and nutrition. Due to its 15 16 properties, zeolites can absorb therapeutically active proteins and release them under physiological conditions. In this study we tested the clinoptilolite, chabazite and natrolite ability 17 to be loaded by antitumor ribonuclease binase and the cytotoxicity of the obtained complexes. 18 19 We found the optimal conditions for binase loading into zeolites and established the dynamic of its release. Cytotoxic effects of zeolite-binase complexes towards colorectal cancer CaCo2 cells 20 were characterized after 24h and 48h of incubation with cells using MTT-test. Zeolites were 21 toxic by itselfs and reduced cells viability by 30% (clinoptilolite), 40% (chabazite) and 70% 22 (natrolite) after 48 h of incubation. Binase complexes with clinoptilolite as well as chabazite 23 always demonstrated enhanced toxicity (up to 57% and 60% for clinoptilolite and chabazite, 24 respectively) in comparison with binase and zeolites separately. Our results contribute to the 25 perspective development of binase-based complexes for therapy of colorectal cancer for or the 26 treatment of malignant skin neoplasms where the complexes can be used in pasty form. 27

### Introduction 28

29 There are about 40 naturally occurring tectosilicate minerals in zeolite group, the most commonly mined isometric forms include chabazite and clinoptilolite, the fibrous form is mainly 30 represented by natrolite. Chemical differentiation of zeolites is related to the ratio of SiO<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub> 31 and water content. Zeolites with an Al/Si ratio of 0.20 to 0.40 are leafy, others with an Al/Si ratio 32 up to 0.50 are isometric or mostly isometric, and with an Al/Si ratio of 0.60-1.00 are 33 34 predominantly fibrous. The crystalline structure of zeolites is formed by tetrahedral  $SiO_{2/4}$  and AlO<sub>2/4</sub> groups, united into a three-dimensional framework pierced by cavities and channels which 35 size is 0,2 - 1,5 nm. The internal cavities and the channels are filled with molecules of water. The 36 37 open frame-cavity structure of zeolites has a negative charge, which is compensated by counterions (metal, ammonium, alkylammonium and other cations). 38

Zeolites are capable to exchange cations and reversible dehydrate. Pores in zeolite let small 39 molecules pass through but trap larger ones; that is why they are referred as molecular sieves. 40 Alumina-rich zeolites are attracted to polar molecules such as water, while silica-rich zeolites 41 work better with nonpolar molecules. Advances in material synthesis lead to engineering of 42

hierarchically organized zeolites with multilevel pore architecture which combine unique 43 chemical functionality with efficient molecular transport [Mitchell et al., 2015]. Natural and 44 synthetic zeolites are used as drying agents, as detergents, and in water and air purifiers. Zeolites 45 are also marketed as dietary supplements to treat cancer, diarrhea, autism, herpes, and hangover, 46 and to balance pH and remove heavy metals in the body. In vivo studies, micronized zeolite has 47 been shown to reduce the spread of cancer and increase the effect of the chemotherapy drug 48 doxorubicin [Zarkovic et al., 2003]. Up today, zeolites have not been studied as an anticancer 49 drug in human clinical trials. A review by Memorial Sloan-Kettering Cancer Center concluded 50 that none of the benefits seen in animals occurs in humans. (https://www.mskcc.org/cancer-51 care/integrative-medicine/herbs/zeolite). However, different zeolite forms must be distinguished: 52 fibrous mordenite is not allowed for medical use, erionite inhalation toxicity is associated with 53 high incidence of malignant mesothelioma [Elmore, 2003; de Assis et al., 2014]. At the same 54 55 time, TMAZ®, a natural isometric zeolite clinoptilolite with enhanced physicochemical properties, is the basis of the dietary supplements Megamin® and Lycopenomin® ("Tribo 56 57 Ming", Croatia), which have demonstrated antioxidant activity in humans. Litovit® ("Nov", 58 Novosibirsk, Russia) that removes heavy metals and has radioprotective properties is also manufactured on the basis of clinoptilolite. The composition synthesized from naturally 59 occurring non-toxic zeolites was patented in US against buccal mucosa and lung squamous 60 epithelial cell cancers [Kaufman, 2001]. 61

Taken together, this data indicates that adsorptive and ion-exchange properties of some zeolites 62 could be applied in medical practice. In our study, several zeolites allocated as possible 63 candidates for loading of anticancer therapeutics. We tested isometric clinoptilolite and chabazite 64 ability to absorb therapeutic protein and realize it, in comparison to this ability of fibrous 65 natrolite. We chose binase (ribonuclease (RNase) from *Bacillus pumilus*) as a therapeutic protein. 66 RNases are potential antitumor drugs due to their cytotoxicity and due to their influence at some 67 68 tumor cells functions. RNases have demonstrated the ability to overcome multidrug resistance and to enhance the cytotoxicity of a variety of anticancer agents [Suri et al., 2007; Zelenikhin et 69 al., 2016]. Binase triggers apoptotic response in cancer cells expressing RAS oncogene which is 70 71 mutated in a large percentage of prevalent and deadly malignancies [Ilinskaya et al., 2001; 72 Cabrera-Fuentes et al., 2012; 2013]. Other microbial RNases, cationic mutants of RNAse Sa, for example, possess similar selective activity to oncotransformed cells [Ilinskaya et al., 2002]. The 73 74 specific antitumor effect of binase towards RAS-transformed cells is due to its direct binding of RAS protein and inhibition of downstream signaling [Ilinskaya et al., 2016]. The expression of 75 oncogenes, in particular, AML1-ETO and kit, was shown to determine the selective sensitivity of 76 77 cells to the binase action. Moreover, the anti-metastatic effect of binase was demonstrated in animal models. Binase at doses of 0.1-1 mg/kg, which produced effective suppression of tumor 78 growth and metastasis, showed positive effect on the liver of tumor-bearing mice expressed in a 79 80 significant reduction of the liver parenchyma destructive changes and return to the normal level the liver regenerative potential [Sen'kova et al., 2014]. Thus, this bacterial RNase can be 81 considered a perspective antitumor agent because of its targeted activity toward certain 82 oncogenes expressing cancer cells. 83

The present study is aimed at the search for a biocompatible mineral carrier that allows the safe 84 delivery and long-term action of binase needed for treatment of ras-expressing malignances, 85 especially colorectal cancer. The delivery of proteins to the intestine is known to be complicated 86 87 by their degradation in digestive tract with subsequent loss of therapeutic activity. Therefore, the prolonged release of antitumor agents from composite pills or rectal suppositories can provide 88 certain advantages. Similarly, these advantages are inherent in therapeutic application of pasty 89 form for the treatment of malignant skin neoplasms. Here, we found the optimal conditions for 90 binase loading into zeolites and established the dynamic of its release. Cytotoxic effects of 91 92 zeolite-binase complexes towards colorectal cancer cells were compared with cytotoxicity of 93 enzyme or zeolite. Our results contribute to the perspective development of binase-based complexes for therapy of colorectal cancer. 94

95

# 96 Materials and methods

*Binase*. The guanyl-preferring RNase from *B. pumilus*, binase (monomer of 12.2 kDa, 109 amino
acid residues, pI 9.5), was isolated from culture fluid of native binase producer as homogenous
protein using the three-step procedure described earlier [Dudkina et al., 2016]. The binase
catalytic activity was determined by measurement of high-polymeric yeast RNA hydrolysis
products according to modified method of Anfinsen [Kolpakov, Il'inskaia, 1999].

102 *Zeolites.* Chabazite [(Ca,Na<sub>2</sub>,K<sub>2</sub>,Mg)Al<sub>2</sub>Si<sub>4</sub>O<sub>12</sub> ×  $6H_2O$ ], the mineral of trigonal syngony, 103 crystallizes in the triclinic crystal system with typically rhombohedral shaped crystals. The 104 crystals are typically twinned, and both contact twinning and penetration twinning may be 105 observed. Crystals of local chabazite up to 5 cm in size have pseudocubic forms, are pale orange 106 with pearly tint and are characterized by a high degree of stoichiometry. In our study we used the 107 samples from Sokolovo-Sarbaisky ore complex, Kazakhstan.

108 <u>Clinoptilolite</u> [(Na,K,Ca)<sub>2-3</sub>Al<sub>3</sub>(Al,Si)<sub>2</sub>Si<sub>13</sub>O<sub>36</sub> × 12H<sub>2</sub>O] forms as white to reddish tabular 109 monoclinic tectosilicate crystals. We used samples from Tatar-Shatrashan deposit of zeolite-110 bearing rocks, Russia. This mineral of the monoclinic syngony exists on the specified deposit in 111 a fine-dispersed state, which is part of a polymineral aggregate consisting of a clayey and 112 siliceous phase (the so-called zeolite-bearing rock). The maximum amount of zeolite in this unit 113 can reach 50%.

114Natrolite  $[Na_2Al_2Si_3O_{10} \times 2H_2O]$  often occurs in compact fibrous aggregates, the fibers having a115divergent or radial arrangement. Natrolite is a mineral of rhombic syngony, in the zones of116metasomatic processing of alkaline igneous rocks of the Kola Peninsula forms large (up to 1 m)117mono- and polycrystalline aggregates of snow-white color with a characteristic silky shine. We118used natrolite from the Khibiny Mountains of the Kola Peninsula, Russia.

Loading/unloading procedure. Zeolite powder obtained after grinding in an electric mill was 119 120 treated with concentrated filtered hydrochloric acid HCl to remove various impurities, washed with MQ-water and dried using dry heat oven at 160°C. Each sample (5mg) was mixed with 121 binase solution in 96% ethanol (1 mg/ml), thoroughly vortexed (V-1 plus, Biosan, Latvia) until a 122 123 homogeneous suspension, and then sonicated in ice for 5 min, 35 kHz, 130 W (Sapphire, Russia) 124 to disintegrate the aggregates. Afterwards samples were incubated for 2 h with gentle shaking (Mini Rocker-Shaker MR-1, Biosan, Latvia) at 25°C. To estimate the part of non-immobilized 125 126 binase, protein concentration by optical density at 280 nm and RNase catalytic activity of the supernatant obtained after centrifugation (5 min, 4300 g, Eppendorf 5415R, Germany) were 127 measured. The sediments were dried at 50°C and stored at room temperature. To analyze the 128 129 release of immobilized enzyme, the samples were suspended in MQ-water and incubated at room temperature for 2 h, 4 h or 6 h. After centrifugation, the binase catalytic activity and protein 130 concentrations were measured in the supernatant. 131

132 *Cell cultures.* Colon adenocarcinoma cells (Caco2) were obtained from Russian cell culture 133 collection (Saint-Petersburg, Russia). Cells were grown in RPMI 1640 medium supplemented 134 with penicillin (100 U/mL), streptomycin (100 U/mL), 2mM glutamine (Sigma-Aldridge, USA) 135 and 10% fetal bovine serum (HyClone, USA) at 37°C in a humidified atmosphere with 5% CO<sub>2</sub>. 136 Cells were seeded into 96-well plates and grown 12 h; then tested samples dissolved in fresh 137 medium were added into plates. After 24 h and 48 h of incubation the MTT assay was 138 performed.

139 MTT-assay. Cell viability was measured according to mitochondrial dehydrogenase activity tested by standard procedure based on the reduction of MTT tetrazolium dye. Cells ( $10^4$  per well 140 in 96-well plate, CELLTREAT Scientific Products, USA) have grown overnight, then cultural 141 142 fluid was discarded and fresh medium with test samples (or with an equivalent volume of water for negative control) was added. After 24 and 48 hours culture medium was replaced with 143 dimethyl sulfoxide (Sigma-Aldrich, USA) to dissolve formazan crystals, when absorption was 144 145 measured at 570 nm (xMark, Bio-Rad, USA). As a positive control inducing cell death 1% Triton was used. 146

147 Transmission Electron Microscopy. Zeolite samples with 96% ethanol solution were sonicated 148 during 10 min, 35 kHz, 130 W (Sapphire, Russia) to disintegrate the aggregates. A droplet of 149 diluted zeolite samples was placed onto carbon-coated grids and left to evaporate. Specimens 150 were inspected using a Hitachi HT7700 Exalens transmission electron microscope (Hitachi 151 High-Tech Science Corporation, Japan) at resolution 1,4 Å. TEM bright field images were 152 recorded at 100 kV accelerating voltage using a AMT XR-81 CCD camera (3296×2742, 8 153 megapixel, 5,5 mm pixel size).

154 *Statistics.* Statistical data analysis and plotting were performed by means of GraphPad Prism6 155 software (United States). The statistically significant level was taken as  $p \le 0.05$ .

156

# 157 **Results**

158 Chabazite is morphologically represented by small oval or pseudocubic particles with a diameter 159 about 100~200 nm aggregated into regular round-shaped particles ( $\emptyset \sim 2\mu m$ ). The same small 160 particles are typical for clinoptilolite, but they form amorphous structures of different size. Small 161 particles of natrolite are partially leafed or polygonal forms (Figure 1).

162 Finely crushed samples of three different zeolites in the form of micrometer particles were used for binase immobilization. Initially, during the selection of binase loading conditions we used 163 aqueous solution of enzyme but the measurement of unloaded protein concentration and RNase 164 165 activity of the supernatant showed the lack of enzyme immobilization. Therefore we used 96% ethanol to solve the enzyme before loading. RNase activity in ethanol solution was almost the 166 same as in water,  $1,116\pm0,013\times10^6$  units/mg and  $1,588\pm0,020\times10^6$  units/mg, correspondingly. 167 More than 80% of the protein was found to adsorbe on all zeolites, whereby residual catalytic 168 activity measured in supernatant was very low. The best results were obtained with chabazite 169 170 (Table 1). Full release of binase from chabazite takes 6 h, for clinoptilolite this time period is 4 h. Natrolite kept residual amount of protein more than 6 h. The main part of protein (more that 80% 171 172 of immobilized one) released from all three zeolites was found in solution already after 2h of incubation (Table 2). RNase activity of released binase was comparable to the activity of pure 173 binase in water. Staying in natrolite reduced the catalytic activity of the enzyme released after 2h 174 175 up to 57%. This effect disappeared after 4h of incubation. Opposite, staying inside chabazite 176 slightly activated the binase catalytic activity (Table 2).

The cytotoxicity of pure zeolites and zeolites loaded with binase was studied on human colon 177 178 adenocarcinoma cell line Caco2. Each type of zeolites (300 µg/ml) was examined on possible toxic effects after 24h and 48h of incubation with growing cells (Figure 2). After addition of the 179 same amount of water used for zeolite suspension the cells viability reduced on 18% compared 180 to growth without any supplements. Pure chabazite inhibited cell viability by less than 40% 181 during all time of cultivation, clinoptilolite showed inhibitory effect of approximately 50% after 182 24h decreased after 48h up to 30%. Natrolite was more toxic, its inhibitory effect increased from 183 184 30% at 24h to 70% at 48h of cell growth.

Pure binase at concentration 100  $\mu$ g/ml reduced cell viability by 60% only after 48h of incubation. Higher concentration (300  $\mu$ g/ml) affected cell viability already after 24h (inhibition reached 40%), after 48h inhibitory effect was 58% (Figure 2). Binase immobilized in natrolite or clinoptilolite increased their toxicity during 24h, then this increase for natrolite, but not for clinoptilolite, was abolished. Complexes of clinoptilolite as well as chabazite with binase always demonstrated enhanced toxicity in comparison with binase and zeolites separately (Figure 2).

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# 192 Discussion

193 Binase possesses selective toxicity toward certain tumor cells *in vitro* [Makeeva et al., 2017;

Mitkevich et al, 2011, 2013, 2015; Zelenikhin et al, 2016, 2016a ] and *in vivo* [Mironova et al., 2014; Sen'kova et al., 2014].

196 The expression of certain oncogenes (*ras, kit, AML1-ETO*) is a marker of tumor cells 197 susceptibility to binase apoptogenic action. In some cases RNases catalytic activity may be an 198 important factor for their cytotoxicity manifestation [Makarov, Ilinskaya, 2003; Ilinskaya,

Vamvakas, 1997]. However, the sensitivity of malignant breast cancer cells to binase apoptosis 199 200 inducing effect was not shown to correlate with the level of cellular RNA catalytic degradation [Zelenikhin et al, 2016]. This effect was also demonstrated for oncogene kit transformed cells 201 [Mitkevich et al., 2010]. Using quantitative RT-PCR with RNA samples isolated from the 202 binase-treated transgenic myeloid progenitor cells FDC-P1-N822K expressing the activated kit-203 204 oncogene (mutation Asn822Lys), we have found that the amount of mRNA of the kit oncogene 205 gene was reduced by half. This means that binase effect to tumor cells is specific and is determined by presence of cells specific molecular targets, which can be certain RNA as well as 206 proteins, in particular, RAS [Ilinskaya et al., 2016]. 207

Therefore, its delivery and prolonged action could have benefits during application against cancer.

Zeolites are a group of calcium and sodium aqueous aluminosilicates similar in composition and 210 properties. In the gut, these silicates could act as adsorbents, catalysts, detergents or anti-211 diarrheic agents to their absorption potential and ion-exchanger properties. Zeolites themselves 212 213 are widely used in agriculture as adsorbents. In animals, zeolite supplementation of feed resulted 214 in a reduction in number of poultry pathogens without damaging the beneficial bacteria [Prasai et al., 2017]. Dietary administration of small particle size clinoptilolite can effectively reduce 215 concentration of aflatoxins in dairy cattle milk [Katsoulos et al., 2016]. So, the detoxificant role 216 of zeolites is already evident in agro and in zoothecnical fields [for review see Laurino C, 217 Palmieri B., 2015]. We started our study from two simple approvals. First, clinoptilolite 218 application in medicine is allowed, preparations "Tribo Ming" (Croatia), and "Nov" (Russia) 219 based on this zeolite are available for purchase in pharmacies. Natural clinoptilolite with 220 enhanced physicochemical properties is the basis of the dietary supplements Megamin and 221 Lycopenomin, which have demonstrated antioxidant activity in humans [Ivkovic et al., 2004]. 222

We have also studied the possibility of other zeolites, chabazite and natrolite to serve as carriers 223 224 for binase. Chabazite was studied previously as an agent for wastewater purification [Lee et al., 2016; Montégut et al., 2016], natrolite was described as an environmentally benign catalyst 225 [Nasrollahzadeh et al., 2017]. Our results have shown that all three zeolites used at this study 226 227 have the possibility to absorb binase, an antitumor bacterial protein. The zeolites crystalline 228 structure is formed by tetrahedral SiO<sub>2/4</sub> and AlO<sub>2/4</sub>, groups, joined by common vertices into three-dimensional framework, penetrated by cavities and channels 2-15 Å in size. The surface of 229 230 zeolites has a negative charge, compensated by counterions (metal cations, ammonium, and other ions) and water molecules. Washing the zeolites with acid allowed us to get rid of carbonate 231 impurities, and the subsequent washing with water and alcohol removed counterions and 232 233 released the negative charge necessary for sorption of cationic binase (PI 9.5) due to electrostatic interactions. 234

We found the conditions suitable for loading more than 80% of protein from ethanol solution 235 236 during 2 h with gentle shaking by room temperature. Natrolite demonstrated slowly decreasing 237 absorption ability compared to clinoptilolite and chabazite. It probably could be connected to its fibrous nature (Table 1). On the other hand, binase was released from natrolite more slowly than 238 from clinoptilolite and chabazite and did not reached 100% output during 6 h. (Table 2). It could 239 be a positive fact for prolonged binase action, but natrolite itself possessed cytotoxicity 240 241 increasing along the time of incubation with the cells up to high value about 40% with the same cytotoxicity value as pure binase. Therefore, clinoptilolite and chabazite have some preferences 242 243 for use as binase carriers. Complex of clinoptilolite with binase induced the cell death comparable to pure binse after 48 h, but during the first 24 h of incubation the release of binase 244 from clinoptilolite induced higher cytotoxicity as pure binase (Figure 2). This data are in 245 accordance with previously obtained results about the capacity of clinoptilolite to be useful in 246 medicine. Zeolite-containing mixture (Hydryeast®) maintaining mucosal immune homeostasis 247 and epithelial integrity, is known to have a suppressive effect on colitis [Lyu et al., 2017]. In 248 249 humans, zeolite supplementation exerted beneficial effects on intestinal wall integrity and accompanied by mild anti-inflammatory effects in aerobically trained subjects [Lamprecht et al., 250

2015]. Treatment of cancer-bearing mice and dogs with micronized zeolite clinoptilolite led to improvement of the overall health status, prolongation of life span and decrease of tumor size in some cases. Combined treatment with doxorubicin and clinoptilolite resulted in strong reduction of the pulmonary metastasis count increasing anticancer effects of doxorubicin [Zarkovic et al., 2003]. Clinoptiolite is also used in water filters, to soil improvement, wastewater treatment and remidiation, in veterinary medicine (in gastrointestinal tract treatment). Chabazit does not have such widespread use.

So, our results could rise especial interest concerning a binase with chabazite complex. First of all, this complex was always more cytotoxic towards Caco2 cells then chabazite or binase themselves. Then, chabazite has low cytotoxicity. Finally, binase release from chabazite is timedependent (Figure 2). Moreover, the catalytic activity of binase was slightly stimulated during staying inside of chabazite (Table 2) possibly due to interaction with cations released from this carrier. It means that chabazite-binase complex could be a perspective anticancer agent.

Binase cytotoxicity has grown with concentration increasing during 24 h of incubation. At 48 h 264 265 of incubation, the difference in cytotoxicity of 100 and 300  $\mu$ g / ml binase was not significant 266 (Figure 2). This could be probably caused by the fact that absorption of binase by cells occured rather quickly, especially in first hours, and reached a practical maximum at 6 h. At this time (6 267 h) we previously described a permeability peak for trypan blue-labeled albumin macromolecule 268 269 across cell membrane of cancer lung epithelial cell monolayers treated with RNase [Cabrera-Fuentes et al., 2013]. Probably, during prolonged incubation, binase adsorption slows down, 270 which leads to cytotoxicity of the two used concentrations differences leveling at 48 h 271 incubation. Over time, we observed increasing toxicity of natrolite, which formed the needle-272 shaped fibrous aggregates, damaging cells. Therefore, natrolit cannot be recommended as a 273 274 carrier of potential therapeutic proteins.

- Now, application of zeolites as materials for various therapeutic substances delivery include 275 276 antitumor ones is intensively studied. The composition synthesized from naturally occurring non-toxic zeolites had a 100% kill rate within 72 hours against buccal mucosa and lung 277 squamous epithelial cell cancers and was non-toxic to healthy human cells [Kaufman et al., 278 279 2001]. Zeolite-based nanoparticles used in generating time-controlled release of 5-fluorouracil 280 from zeolite preparations showed anti-cancer effect towards Caco-2 monolayers [Spanakis et al., 2016]. Earlier, we have demonstrated that binase-halloysite complex doubled anticancer 281 282 efficiency of binase due to its perfect absorption by cells and longer release reducing the viability of human colon adenocarcinoma cells Colo320 by 60% [Khodzhaeva et al., 2017]. The same 283 level of toxicity toward human adenocarcinoma Caco2 cells was obtained for chabazite-binase 284 285 complex. At the first time, we have shown that not only clinoptiolite but also chabazite could be used as carriers for new antitumor agents inducing prolonged cytotoxicity towards cancer cells. 286 Moreover, chabazite could help to counteract oxidative stress in apparently healthy subjects 287 288 exposed to different oxidative stress risk factors affecting the levels of different antioxidant 289 enzymes (gluthatione peroxidase, superoxide dismutase, gluthatione reductase [Dogliotti et al., 2012]. Our results indicates that (a) the toxicity of chabazite is insignificant in magnitude and 290 does not increase with time; (b) its complex with binase exhibits cytotoxicity increasing with 291 time due to release of binase from the complex; (c) the level of complex toxicity is slightly 292 293 higher in comparison with pure binase. These facts could open the prospect of using chabazite as a carrier for potential therapeutics proteins. 294
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# 296 Acknowledgments

297 Authors thank Mr. Y. Osin for technical help with TEM images.

Funding. The study was performed within the Russian Government Program of Competitive
Growth of Kazan Federal University and was supported by the Russian Foundation for Basic
Research (project no. 17-00-00060).

- 301
- 302 <u>Author Contributions</u>

303 VK, OL and OI planned experiments. VK and PZ performed experiments. OL and OI analyzed
 304 data. VK and OI wrote the paper.

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# 306 <u>Conflict of Interest Statement</u>

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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review

## 421

- Figure 1. TEM images of three different zeolites (Chabazite A, Clinoptilolite- B, Natrolite C) grinded in an electric mill up to particles of micrometer size. bar =  $200 \,\mu$ m.
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Figure 2. Cytotoxicity of binase-loaded zeolites (300  $\mu$ g/ml), pure zeolites (300  $\mu$ g/ml) and pure

binase at two concentrations, 100  $\mu$ g/ml and 300  $\mu$ g/ml, toward human colon adenocarcinoma

427 Caco2 cells. Data represent mean  $\pm$  SEM of three independent experiments; \*, \*\*, \*\*\*, \*\*\*\* - P < 428 0,05, 0,03, 0,014, 0,01, correspondingly vs. negative control obtained by adding water volume

- 429 equal to volume of zeolite suspension and binase solution; ns, non-significant. Cell viability
- 430 without any additives was taken as 100%...
- 431

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435	The amount of binase loaded onto zeolite and silica samples from ethanol solution <sup>a</sup> .				
	Zeolite sample	Loaded enzyme, %			
		Measured by protein c	concentration	Measured b	by catalytic activity
	Chabazite	86,9 ± 1,	9		$100 \pm 0.8$
	Clinoptilolite	$85,4 \pm 0,$	8	$99 \pm 0,9$	
	Natrolite	$83,9 \pm 1,$	2	$98 \pm 1,2$	
437 438 439 440 441 442	(SmartSpec Plus, Bio-Rad, USA); the catalytic activity was estimated as described in Materials and Methods. The initial concentration of protein (1 mg/ml) and RNase catalytic activity in ethanol solution $(1,1\pm0,013\times10^6 \text{ units/mg})$ used for enzyme loading was taken for 100%. Values are means $\pm$ SD. Experiments were performed in triplicate with five independent replications in each series.				
443 444 445					
446	Table 2.				
447	The amount of binase protein released from zeolite samples into MQ-water and catalytic				
448	activity of the released enzyme.				
	Zeolite sample	Released protein <sup>a</sup> / catalytic activity <sup>b</sup> , %			
	Time of incubation, h	2	4		6
	Chabazite	$90,7\pm0,7$ / 147 $\pm$ 24	93,1 ± 1,0 /1	33 ± 18	$100 \pm 0,4 \text{ / } n^{c}$
	Clinoptilolite	98,0±0,9/115±17	$100 \pm 0.5 / 9$	$97 \pm 19$	$100 \pm 0.6 / n$

<sup>a</sup> The amount of loaded binase was taken for 100% 449

<sup>b</sup> Catalytic activity RNase dissolved in MQ-water ( $1,6\pm0,02\times10^6$  units/mg) was taken for 100%. 450

87,6 ± 0,3 / 112 ± 16

<sup>c</sup> n – Not measured. Values are means  $\pm$  SD. 451

Experiments were performed in triplicate with five independent replications in each series. 452

82,7 ± 0,5 / 57 ± 21

453

Natrolite

 $98,0 \pm 0,4 / n$ 







48 h

